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Workbook112Limiting Reactants.  
Limiting Reactants. At the beginning of  
Chapter 8, a comparison was made  
between solving stoi-  
chiometry  
problems and making turkey

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sandwiches. Look at the sandwich recipe once more: 2 bread slices 2 turkey slices 1 lettuce leaf 1 cheese slice → 1 turkey-and-cheese sandwich If you have 24 slices of turkey, you can make 12 sandwiches at 2 slices per sandwich if you have enough of all the other ingredients.

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## **Skills Worksheet Problem Solving**

Holt ChemFile: Problem-Solving  
Workbook 48 Mole Concept Mole  
Concept Suppose you want to carry out  
a reaction that requires combining one  
atom of ... Yes; the answer has the  
correct units of atoms of boron. Is the  
number of significant figures correct?  
Yes; the mass of boron was given to

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three significant figures. 1 mol B

## **Skills Worksheet Problem Solving**

Holt ChemFile: Problem-Solving

Workbook 100 Stoichiometry Name

Class Date Problem Solving continued

COMPUTE EVALUATE Are the units

correct? Yes; the answer has the correct units of moles  $\text{NH}_3$ . Is the number of

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significant figures correct? Yes; two significant figures is correct because data were given to two significant figures. Is the answer reasonable?

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three significant figures. 1 mol B

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Holt ChemFile: Problem-Solving  
Workbook 9 Conversions Name Class  
Date Problem Solving continued Rewrite  
the given quantity in simple units as  
follows. m<sup>3</sup> m m m COMPUTE 0.50 m<sup>3</sup>

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neon 0.50 (m m m) 10 1 0 m cm 10 1 0  
m cm 10 1 0 m cm 500 000 cm<sup>3</sup> neon  
EVALUATE Are the units correct? Yes;  
cubic centimeters were the units sought.  
Is the answer reasonable?

## **Skills Worksheet Problem Solving**

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Workbook259pH. pH. In 1909, Danish

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biochemist S. P. L Sørensen introduced a system in which acidity was expressed as the negative logarithm of the H concentration. In this way, the acidity of a solution having H concentration of  $10^{-5}M$  would have a value of 5.

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Workbook 112 Limiting Reactants  
Limiting Reactants At the beginning of Chapter 8, a comparison was made between solving stoi- ... Is the answer reasonable? Yes; you can see that 6.3 mol HCl is more than is needed to react with 2.8 mol Ca(OH)<sub>2</sub>. Practice 1.

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## **Ed W. Clark High School**

$N_2(g) + 3H_2(g) \rightarrow 2NH_3(g)$  In the manufacture of ammonia, it is nearly impossible to produce 2 mol (34.08 g) of  $NH_3$  from the simple reaction of 1 mol (28.02 g) of  $N_2$  and 3 mol (6.06 g) of  $H_2$ , because some ammonia molecules begin breaking down into  $N_2$  and  $H_2$  molecules as soon as they are formed.



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Holt Chemistry File: Mini-Guide to  
Problem Solving 1st Edition by  
RINEHART AND WINSTON HOLT (Author)  
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step-by-step solutions and a large  
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difficulty levels with the answers given

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which you determine the concentration  
of a solution by measuring what volume  
of that solution is needed to react  
completely with a standard solution of  
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