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Percentage composition can also be used to determine the mass of a certain element that is contained in any mass of a compound. In the previous sample problem, it was found that the percent composition of dichlorine heptoxide is $(38.76\% \text{ Cl})$ and $(61.24\% \text{ O})$. Suppose that you needed to know the masses of chlorine and oxygen ...

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Percentage Composition Worksheet - Answer Key Give the % composition of all elements in these compounds. Show all work! ____ 1) ammonium sulfite % N 24.1 % H 6.9 % S 27.6 % O 41.3 ____ 2) aluminum acetate % Al 13.2 % C 35.3

Percentage Composition Worksheet

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PERCENTAGE COMPOSITION Name ine the percentage composition of each of the compounds below. Det 1. Mnoa 2, 4. (NH4)3PO4 (7 16e 77563 Z Al Solve the following problems. 6. How many grams of oxygen can be produced from the decomposition of 100. g '100 JS/7(1 7 eIf Y/ 7. How much iron can be recovered f om 25.0 g of Fe O ?

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