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Textbook pages: Chapter 12. Key Terms: stoichiometry. mole-mole problems. mass-mass problems. mass-volume problems. volume-volume problems. particle -particle problems. expected yield. actual yield. percent yield Directions: Use this information as a general reference tool to guide you through this unit. Don't hesitate to ask your teacher ...

CHAPTER 11: STOICHIOMETRY

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CHAPTER 9 REVIEW Stoichiometry SECTION 3 PROBLEMS Write the answer on the line to the left. Show all your work in the space provided. 1. 88% The actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield. 2. 6.0 mol of N₂ are mixed with 12.0 mol of H

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ANSWER KEY 4 49.3% Rh 23.4 % C 27.3 % N 5. a) PBr 5 b) Zr(BO 3) 2 Zirconium (VI) borate 6. a) C 3H 5Cl b) C 6H 10Cl 2 7. empirical: KCO 2 molecular: K 2C 2O 4 name: Potassium oxalate 8. You will be setting up a balanced equation. Using the Law of Conservation of mass (FIRST!) you will be using Mole stoichiometry to determine the balanced equation.

Unit 3 Toombs - cpb-ca-c1.wpmucdn.com

Chapter 3 Review Answers. ... Ch 3 Worksheet Answers. Ch 3 Handout Answers. Key Stoichiometry. AP Quest Ans. Hydration Key . Mole Calculations 1. Mole Calculations 2. Mole Calculations 3 . Mole Mass and Percent Comp . Ch 4: ... Organic and Nuclear Practice Test Key. Organic and Nuclear Worksheet Keys . AP exam keys: AP Part 2 Key 2014 .

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